

Tandem Wittig Reaction and Cope Rearrangement of 2-Allyl-1,2-dihydroindol-3-ones to 3-Indole Acetates

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Treatment of 2-allyl-1,2-dihydroindol-3-ones **1** with phosphonium ylides **2** in refluxing toluene gives 3-indole acetates **4** in good yields by tandem Wittig reaction and aromatization induced Cope rearrangement.

The Cope rearrangement has found substantial utility in the methodology of synthetic organic chemistry.¹ The rearrangement of unsubstituted hexa-1,5-diene is reversible to form preferentially highly substituted olefins in an equilibrium mixture. In order to completely shift the equilibrium in the desired direction, the reaction has exploited driving forces such as conjugation,^{2,3} strain⁴ or irreversible conversion⁵ of one diene to a more stable product. However, there has been little use of aromatization⁶ (*i.e.* the incorporation of a double bond of hexa-1,5-diene into an aromatic ring) to effect the Cope rearrangement, because of the difficulty of constructing the starting hexa-1,5-diene system. We here describe tandem Wittig olefination and aromatization induced Cope rearrangement of

indolin-3-ones, which provides a new method for the synthesis of biologically interesting 3-indole acetic acid derivatives.⁷

The starting 2-allyl-1,2-dihydroindol-3-ones **1-c** were readily available by our previously described methods.⁸ When the allylindol-3-one **1a** was treated with the phosphonium ylide **2a** in refluxing toluene for 5 h, the Wittig olefination followed by the Cope rearrangement of an intermediary 3-alkylidene-1,2-dihydroindole **3a** occurred smoothly to give 2-(indol-3-yl)pent-4-enoate **4a**[†] in 87% yield. Similar reaction of indol-3-ones **1a-c** with ylides **2a-f** afforded the corresponding 3-indole derivatives **4b-g** in good yields[‡] (Table 1).

As can be seen from Table 1, either the bulkiness of the substituents in the ylide **2c** (entries 1 vs. 3) and 1,1-dimethylallyl group in the indol-3-one **1b** (entries 1 vs. 5) or the nucleophilicities⁹ of the ylides (**2a, e, f**) (entries 5-7) affected the reaction, especially the initial Wittig reaction step. In the case of **1c**, the initial Wittig reaction proceeded smoothly, but the Cope rearrangement was slower due to the bulkiness of the 3,3-dimethylallyl group of **1c**. In all the cases examined, the Cope rearrangement of 1,5-diene **3** caused both aromatization and deconjugation of the olefins with ester, keto or cyano groups. This indicates the superiority of aromatization over conjugation as the well-known driving force in Cope rearrangement. Although it was found in our previous work¹⁰ that the Wittig reaction of 2-unsubstituted and 2-alkylated 1,2-dihydroindol-3-ones followed by a formal [1,3] hydrogen shift afforded 3-alkylindoles, the formation of such a product **5** due to hydrogen shift was not observed except for **1c**. Therefore, in the present reaction, [3,3]-sigmatropic rearrangement of the intermediate **3** was preferable to the [1,3]-sigmatropic shift.

Further applications and studies on the stereochemistries of these tandem reactions are in progress.

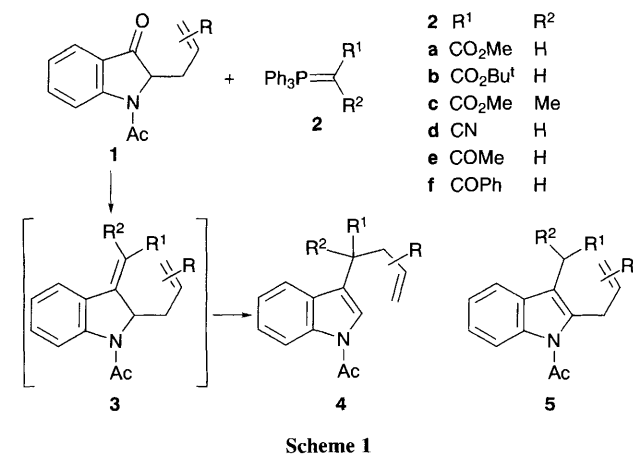


Table 1 Tandem Wittig reaction and Cope rearrangement of 2-allyl-1,2-dihydroindol-3-ones

Entry	Indol-3-one 1	Ylide 2	Reaction ^a time/h	Product 4	[Yield (%) ^b]
1		2a	5		4a R ¹ = CO ₂ Me, R ² = H (87%)
2		2b	6		4b R ¹ = CO ₂ Bu ^t , R ² = H (78%)
3		2c	10		4c R ¹ = CO ₂ Me, R ² = Me (81%)
4		2d	40		4d R ¹ = CN, R ² = H (61%)
5		2a	14		4e R ¹ = CO ₂ Me (58%)
6		2e	62		4f R ¹ = COMe (58%) ^c
7		2f	72		4g R ¹ = COPh (32%) ^c
8		2a	20		4h (34%) ^d

^a The reaction was run in refluxing toluene. ^b Isolated yield. ^c Recovered **1b** (21%, entry 6; 61%, entry 7). ^d The Wittig product **3h** (50%) and 2-(2,3-dimethylallyl)indole acetate **5h** (16%) were also obtained.

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Footnotes

† Selected spectroscopic data for **4a**: IR (CHCl₃) ν/cm^{-1} 1702, 1453, 1384, 1353, 1332. ¹H NMR (270 MHz, CDCl₃, J/Hz) δ 2.63 (s, 3H), 2.70 (dddd, J 1.3, 6.6, 8.3, 14.2, 1H), 2.87 (dddd, J 1.3, 6.6, 7.3, 14.2, 1H), 3.70 (s, 3H), 3.93 (dd, J 7.3, 8.3, 1H), 5.06 (ddd, J 1.3, 1.7, 10.2, 1H), 5.12 (ddd, J 1.3, 1.7, 17.2, 1H), 5.80 (ddt, J 6.6, 10.2, 17.2, 1H), 7.30 (dt, J 1.0, 7.6, 1H), 7.36 (ddd, J 1.0, 7.6, 7.9, 1H), 7.40 (s, 1H), 7.38 (dt, J 1.0, 7.6, 1H), 8.43 (br d, J 7.9 Hz, 1H). HRMS m/z 271.1200 (calc. C₁₆H₁₇NO₃ 271.1208).

‡ Satisfactory analytical and spectral data were obtained for all compounds.

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